

2-Amino-1-phenylimidazole hydrochloride. Obtained from 1-phenyl-2-sodioimidazole. Mp 206–206.5° C. Yield 43.5%. Found, %: C 55.23; H 4.75; Cl 18.20; N 21.80. Calculated for $C_9H_9N_3 \cdot HCl$, %: C 55.25; H 5.16; Cl 18.11; N 21.43.

2-Amino-1-phenylimidazole. Mp 125–126° C (from hexane). Found, %: C 67.70; H 5.83; N 26.45. Calculated for $C_9H_9N_3$, %: C 67.90; H 5.69; N 26.40.

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Rostov-on-Don State University

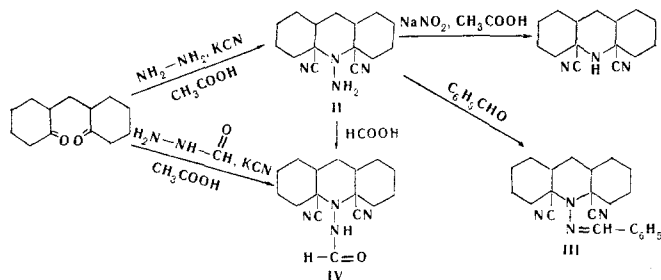
CYCLOHYDRAZINOCYANATION OF 2,2'-METHYLENE-BIS-CYCLOHEXANONE

V. A. Kaminskii, V. K. Gamov, and M. N. Tilichenko

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With 2,2'-methylene-bis-cyclohexanone (I) as an example, we have established that 1,5-diketones of this type are capable of undergoing a cyclohydrazinocyanation reaction analogous to the cycloaminocyanation reaction [1]. The reaction takes place exceptionally readily when the given ketone is mixed with KCN and hydrazine in dilute acetic acid; this gives a 90% yield of N-amino-11,14-dicyanoperhydroacridine (II). 1,5-Diketones of the aliphatic series have previously been subjected to reaction with hydrazine and hydrocyanic acid, but only in an aqueous medium, which gives the corresponding N-amino-2,6-dicyanopiperidines in low yields [2,3].



On reaction with benzaldehyde in propanol with the addition of acetic acid anhydride, II forms the N-benzylidene derivative III. The structure of II is shown definitively by the fact that on treatment with sodium nitrite in acetic acid it undergoes deamination and is converted into the known 11,14-dicyanoperhydroacridine [1].

On being boiled with 85% HCOOH, II gives the N-formyl derivative IV. The same compound is also formed by the reaction of I, KCN, and formhydrazide in dilute acetic acid.

II. Colorless crystals, mp 210–211° C (benzene–dioxane 1:2). Found, %: C 69.55; H 8.76; N 21.45. Calculated for $C_{15}H_{22}N_4$, %: C 69.77; H 8.53; N 21.70. IR spectrum (in CCl_4), cm^{-1} : 3380 (NH), 2220 (C≡N). III. Light yellow crystals, mp 225–228° C (ethanol). Found, %: C 76.22; H 7.69; N 16.31. Calculated for $C_{22}H_{26}N_4$, %: C 76.30; H 7.51; N 16.18. IV. Colorless crystals, mp 206–208° C (from 40% ethanol). Found, %: C 67.29; H 8.12; N 19.87. Calculated for $C_{16}H_{22}N_4$, %: C 67.13; H 7.69; N 19.58. IR spectrum (in KBr), cm^{-1} : 1690 (amide C=O), 2230 (C≡N).

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Far Eastern State University,
Vladivostok